## Concentration-Dependent Emission Spectra in Cr<sup>3+</sup>:GGG Transparent Nanoceramics

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In recent years Cr-doped garnets attracted a lot of attention due to their unique properties. Interpretation of spectroscopic properties of  $Cr^{3+}$  ions in garnets is difficult due to the variation of the crystal field strength in a large range resulting in the presence of  $Cr^{3+}$  ions in octahedral sites with different local crystal field strengths. This is relevant also for nanostructured materials such as  $Cr^{3+}$ -doped nanopowders. Despite the large number of works, a lot of questions remain concerning the luminescence properties of  $Cr^{3+}$  ions in nanomaterials.

In the present work we report the studies of optical properties of Cr<sup>3+</sup>:GGG nanoceramics synthesized by high isostatic pressure (HIP) method at relatively low temperature. Cr:GGG nanoceramics doped with 0.1 at.%, 0.2 at.%, and 0.3 at.% of chromium ions were obtained.

Synthesized nanoceramics consist of crystallites with the average sizes of 17 nm and the average lattice parameters ~12.40 Å. High temperature emission spectra show the presence of at least four different  $CrO_6$  optical active centers in Cr:GGG nanoceramics. The presence of satellites of R-lines confirms uniform distribution of local crystal fields around different chromium ions. The change in the concentration of  $Cr^{3+}$  ions do not influence the room temperature optical properties of Cr:GGG nanoceramics. Excitation and emission spectra, and lifetimes were the same for the investigated samples. No difference in the calculated Racah parameters was detected for different  $Cr^{3+}$  concentrations.

The low temperature emission spectra depend on the chromium content and the increase in  $Cr^{3+}$  concentration caused an increase in the ratio of broadband  ${}^{4}T_{2g}({}^{4}F)$  emission intensity to overall  $Cr^{3+}$  emission. At the same time no difference was detected in the peak position and width of  ${}^{4}T_{2g}({}^{4}F) \rightarrow {}^{4}A_{2g}({}^{4}F)$  broadband emission. The difference in the low temperature emission spectra was explained by the energy transfer between chromium ions in the sites with different local crystal field strength.  $Cr^{3+}$  ions in the low crystal field site have higher probability of emitting photons rather than transferring the excitation energy to the nearby  $Cr^{3+}$  ions. The increase in concentration of  $Cr^{3+}$  ions caused an increase in the energy transfer between  $Cr^{3+}$  ions thus increasing the ratio of emission intensity of  $Cr^{3+}$  ions in low crystal field to the total luminescence.

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